# Important roles of the conserved linker-KKS in human neuronal growth inhibitory factor

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**Abstract** Metallothinein-3 (MT3), also named neuronal growth inhibitory factor (GIF), is attractive by its distinct neuronal growth inhibitory activity, which is not shared by other MT isoforms. The polypeptide chain of GIF is folded into two individual domains, which are connected by a highly conserved linker, KKS. In order to figure out the significance of the conserved segment, we constructed several mutants of human GIF (hGIF), including the K31/32A mutant, the K31/32E mutant and the KKS-SP mutant by site-directed mutagenesis. pH titration and DTNB reaction exhibited that all the three mutations made the β-domain lower in stability and looser. More significantly, change of KKS to SP also altered the general backbone conformation and metal–thiolate

cluster geometry. Notably, bioassay results showed that the bioactivity of the K31/32A mutant and the K31/32E mutant decreased obviously, while the KKS-SP mutant lost inhibitory activity completely. Based on these results, we proposed that the KKS linker was a crucial factor in modulating the stability and the solvent accessibility of the  $Cd_3S_9$  cluster in the  $\beta$ -domain through domain–domain interactions, thus was indispensable to the biological activity of hGIF.

**Keywords** Metallothionein · Neuronal growth inhibitory factor · Linker · Mutation · Cell culture

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### **Abbreviations**

MT Metallothionein **GIF** Growth inhibitory factor AD Alzheimer's disease PD Parkinson's disease **CNS** Central nervous system Complimentary DNA cDNA **IPTG** Isopropyl  $\beta$ -D-thiogalactoside **DTNB** 5,5'-Dithiobis-(1-nitrobenzoic acid)

PCR Polymerase chain reaction

DTT Dithithreitol

ESI-MS Electrospray ionization mass spectrometry

PSI Pounds per square inch
UV-vis Ultraviolet-visible
CD Circular dichroism

LMCT Ligand-to-metal charge transfer



NMR Nuclear magnetic resonance

MTT 3-(4,5)-dimethylthiahiazo (-2-y1)-2,5-di-

phenytetrazolium bromide

DMSO Dimethyl sulfoxide

## Introduction

Essential transition metal ions, such as copper and zinc, play important roles in neurobiology (Cai et al. 2005). Dysregulated metal metabolism occurs in many neurodegenerative disorders, such as Alzheimer's disease (AD; Adlard and Bush 2006), Parkinson's disease (PD; Rasia et al. 2005) and prion disease (Brown and Kozlowski 2004) etc. Brain has a regulatory network to achieve metal ions homeostasis. One component of this network is the protein family called metallothioneins (MTs; Kagi and Schaffer 1988; Margoshes and Vallee 1957).

Metallothioneins are small (~7 kDa), cysteinerich proteins that bind both essential (Zn<sup>2+</sup>, Cu<sup>+</sup>) and toxic (Cd<sup>2+</sup>, Hg<sup>2+</sup>) metal ions (Kagi and Schaffer 1988). Four mammalian MT isoforms (MT1, MT2, MT3 and MT4) have been identified since its discovery. The third isoform, MT3, also named neuronal growth inhibitory factor (GIF), is mainly expressed in the central nervous system (CNS) and possesses neuronal growth inhibitory properties in vitro that distinguish it from the MT1 and MT2 isoforms (Palmiter et al. 1992; Uchida et al. 1991;

Uchida and Tomonaga 1989). As a member of MT family, the amino acid sequence of GIF exhibits about 70% identity with those of MT1/2, including the preserved array of 20 cysteine residues. However, there are two inserts in GIF compared to MT1/2: a threonine (Thr) at position 5 and a glutamate-rich peptide near the C-terminus. In addition, all known GIF sequences contain the conserved <sup>6</sup>CPCP<sup>9</sup> motif which is absent in all other members of MT family (Fig. 1).

Mutational studies have demonstrated that mutation of the unique  $^6\text{CPCP}^9$  motif either to  $^6\text{CSCA}^9$  or  $^6\text{CTCT}^9$  abolishes the bioactivity of GIF (Hasler et al. 2000; Romero-Isart et al. 2002). The results of our simulation clearly showed that the  $^6\text{CPCP}^9$  motif in the  $\beta$ -domain would induce constraints on the human GIF (hGIF) peptide chain and make the N-terminus (residue 1–13 of hGIF) form a particular conformation with two parallel proline residues (Ni et al. 2007). It is believed that such a conformation would function as "a rigid arm providing an interacting surface for protein–protein interactions" (Williamson 1994). The hydroxyl group of Thr5 is also pivotal to perform the bioactivity of GIF (Cai et al. 2006; Romero-Isart et al. 2002).

Since the discovery of GIF in human brain, the gene has been cloned from many mammalian species and the encoded proteins have also been isolated and studied extensively (Chung et al. 2002a; Kobayashi et al. 1993; Palmiter et al. 1992; Pountney et al. 1994; Uchida et al. 1991). However, till now, the structure

α-domain

Fig. 1 Amino acid sequences of mammalian metallothioneins. Polypeptides of all mammalian MTs are folded into two separate domains: the N-terminal β-domain and C-terminal α-domain. The *black parts* indicate linker sequences

MT1a_Human	MDPN-CSCATGGSCTCTGSCKCKECKCNSC <b>KKS</b> CCSCCPMSCAKCAQGCICKGASEKCSCCA
MT1g_Human	${\tt MDPN-CSCAAGVSCTCASSCKCKECKCTSC\textbf{KKS}CCSCCPVGCAKCAQGCICKGASEKCSCCA}$
MT1_Mouse	${\tt MDPN-CSCSTGGSCTCTSSCACKNCKCTSC} \textbf{KKS} {\tt CCSCCPVGCSKCAQGCVCKGAADKCTCCA}$
MT1_Rat	MDPN-CSCSTGGSCTCSSSCGCKDCKCTSC <b>KKS</b> CCSCCPVGCSKCAQGCVCKGASDKCTCCA
MT2a_Human	MDPN-CSCAAGDSCTCAGSCKCKECKCTSC <b>KKS</b> CCSCCPVGCAKCAQGCICKGASDKCSCCA
MT2_Mouse	MDPN-CSCAAGDSCTCAGSCKCKECKCTSC <b>KKS</b> CCSCCPVGCAKCAQGCICKGASDKCSCCA
MT2_Rat	MDPN-CSCATDGSCSCAGSCKCKQCKCTSC <b>KKS</b> CCSCCPVGCAKCSQGCICKEASDKCSCCA
MT4_Human	MDPGECTCMSGGICICGDNCKCTTCSCKTC <b>RKS</b> CCPCCPPGCAKCARGCICKGGSDKCSCCP
MT4_Mouse	MDPRECVCMSGGICMCGDNCKCTTCNCKTCRKSCCPCCPPGCAKCARGCICKGGSDKCSCCP
MT3_Mouse	$\verb MDPETCPCPTGGSCTCSDKCKCKGCKCTNCKKS   CCSCCPAGCKKCAKDCVCKGEEGAKAEAEKCSCCQ  $
MT3_Rat	$\verb MDPETCPCPTGGSCTCSDKCKCKGCKCTNCKKS  CCSCCPAGCEKCAKDCVCKGEEGAK AEKCSCCQ $
MT3_Human	$\verb MDPETCPCPSGGSCTCADSCKCEGCKCTSCKKS  CCSCCPAECEKCAKDCVCKGGEAAEAEAEKCSCCQ $
K31/32A	$\verb MDPETCPCPSGGSCTCADSCKCEGCKCTSC  \textbf{AAS} \\ \verb CCSCCPAECEKCAKDCVCKGGEAAEAEAEKCSCCQ  \\ \\   \textbf{ABSCCSCCPAECEKCAKDCVCKGGEAAEAEAEKCSCCQ} \\   \textbf{ABSCCSCCPAECEKCAKDCVCKGGEAAEAEAEAEKCSCCQ} \\   ABSCCSCCPAECEKCAKDCVCKGGEAAEAEAEAEAEAEAEAEAEAEAEAEAEAEAEAEAEA$
K31/32E	$\verb MDPETCPCPSGGSCTCADSCKCEGCKCTSCEES  CCSCCPAECEKCAKDCVCKGGEAAEAEAEKCSCCQ $
KKS-SP	MDPETCPCPSGGSCTCADSCKCEGCKCTSC <b>SP-</b> CCSCCPAECEKCAKDCVCKGGEAAEAEAEKCSCCO

**β**-domain



of the intact GIF molecule has not been determined by NMR or X-ray diffraction due to the highly dynamic properties of GIF, especially the  $\beta$ -domain (Faller et al. 1999; Oz et al. 2001; Wang et al. 2006). The spectroscopic characterization revealed that the polypeptide of GIF, like those of recombinant MT1/2, is folded into two domains connected by a short linker and binds seven bivalent metal ions. These metal ions are organized into two individual clusters: a distorted chair three-metal cluster,  $M^{II}_{3}S_{9}$ , in the N-terminal  $\beta$ -domain; and an adamantane-like fourmetal cluster,  $M^{II}_{4}S_{11}$ , in the C-terminal  $\alpha$ -domain (Faller and Vasak 1997; Hasler et al. 1998).

Interestingly, the linker between the two domains is so conserved that it exists as a Lys-Lys-Ser (KKS) segment (Fig. 1) in all mammalian metallothioneins (except MT4, in which the conservative substitution linker RKS appears), which promotes us to explore the possible roles of the conserved linker sequence in structure and function of MTs. In order to figure out the role of the two positively charged lysine residues in neuronal growth inhibitory function of hGIF, we changed these two residues to uncharged residue Ala (the K31/32A mutant) and negatively charged residue Glu (the K31/32E mutant), respectively. In the crustacean MTs, the linker usually exists as XP (X stands for S, A or P) (Narula et al. 1995). In order to find out the functional difference between these two sequences, we changed KKS to SP (the KKS-SP mutant). Then UV-vis spectroscopy, CD spectroscopy, pH titration and DTNB reaction were employed to examine their structure and property differences. Finally, the inhibitory activities of hGIF and its variants were measured by rat neuronal cultures.

## Materials and methods

## Reagents

Fusion expression vector pGEX-4T-2, *Escherichia coli* strain BL21, glutathione Sepharose 4B, Superdex-75 and Sephadex G-25 were purchased from Phamacia Biotech (Uppsala, Sweden). T4 DNA ligase, dNTP and restriction enzymes (BamHI and EcoRI) were purchased from New England Biolabs (Ipswich, MA, USA). Pfu DNA polymerase, cell culture reagents, isopropyl  $\beta$ -D-thiogalactoside (IPTG) and Triton-100

were purchased from Sangon (Shanghai, China). 2, 2'-dithiodipyridine, 5, 5'-Dithiobis-(1-nitrobenzoic acid) (DTNB), bovine thrombin (Catalog No. T4265) and 3-(4,5-dimethylthiahiazo-2-y1)-3,5-di-phenytetrazolium bromide (MTT) were from Sigma (St. Louis, MO, USA). Neurobasal-A medium and B27 serumfree supplements were purchased from GIBCO-BRL (Gaithersburg, MD, USA). The other reagents were of analytical grade.

Cloning, expression and purification of hGIF and its variants

The hGIF gene was obtained by reverse transcription followed by polymerase chain reaction (PCR). The genes of the three mutants were constructed by PCR mutagenesis. Then, each PCR product was digested with *Bam*HI and *Eco*RI restriction enzymes and cloned into vector pGEX-4T-2.

Expression and purification of hGIF and its mutants were carried out as previously described (Yu et al. 2002). The final products were desalted, lyophilized and stored at  $-80^{\circ}$ C.

## Protein reconstitution and characterization

The apo- and fully  $Cd^{2+}$ - or  $Zn^{2+}$ -loaded hGIF and its mutants were generated as previously described (Vasak 1991). The metal-to-protein ratios were determined by measuring the protein concentration via photometric sulphydryl groups (CysSH) quantification upon reaction with 2,2'-dithiodypridine in 0.2 M sodium acetate, 1 mM EDTA (pH 4.0), using an  $\varepsilon_{343} = 7,600 \, \mathrm{M}^{-1} \, \mathrm{cm}^{-1}$  (Grassett and Murray 1967). The metal ion content in each protein was analyzed by flame atomic absorption spectrophotometry (WFX-110, BRAIC, Beijing, China).

Electrospray ionization mass spectrometry (ESI-MS) was used to determine the molecular weights of hGIF and its mutants. Each protein was dissolved in 1% formic acid (v/v) at a concentration of 10 ng ml<sup>-1</sup>. The measurement was carried out on a Bruker Esquire 3000 electrospray ionization mass spectrometer (Bruker Daltonics, Bremen, Germany). The instrument conditions were listed as below: capillary volt, 4 kV; dry gas, 5 l min<sup>-1</sup>; nebulizer gas, 15 PSI (pounds per square inch); infusion flow rate, 3 μl min<sup>-1</sup>.



Ultraviolet-visible and circular dichroism spectra

Ultraviolet-visible (UV-vis) absorption spectra were scanned from 200 to 300 nm on a HP8453 (Hewlett-Packard, Palo Alto, CA, USA) UV-vis spectrophotometer at room temperature using a 1.0 cm quartz cuvette. Circular dichroism (CD) spectra were measured in the range of 200–300 nm on a Jasco (Tokyo, Japan) J-715 spectropolarimeter at room temperature in 10 mM Tris–HCl, 100 mM KCl, pH 8.0.

## pH titration

pH titration was performed as follows:  $6-7 \mu M Cd^{2+}$ -reconstituted proteins were dissolved in 10 mM Tris–HCl, 100 mM KCl, pH 8.0, then titrated with increasing amounts of 1 M HCl (Shaw et al. 1991). The pH was measured directly in the cuvette with a pH microelectrode (SCHOTT-16PH, Germany). The progress of acidification was recorded by UV-vis spectrophotometer. Apparent  $Cd^{2+}$ -binding constants at pH 7.0 of hGIF and its mutants were determined as described previously (Kagi and Vallee 1961; Shaw et al. 1991; Vasak and Kagi 1983; Wang et al. 1994). In all case, the pKa values of cysteine side chains were assumed to be equal to those of rabbit MT isoforms (pKa = 8.9) (Vasak and Kagi 1983).

# Reaction with DTNB

Reactions of  $Cd^{2+}$ -reconstituted hGIF and its mutants with DTNB were studied as described previously (Shaw et al. 1991): 3.5  $\mu$ M protein reacted with 1 mM DTNB in 10 mM Tris–HCl, pH 8.0, containing 100 mM KCl at 25°C. The reaction was monitored at 412 nm ( $\epsilon_{412}=13,600~\text{mol}^{-1}~\text{cm}^{-1}$ ) on a HP8453 UV-vis spectrophotometer at 20 s intervals for 60 min.

## Preparation of rat brain extracts

Rat brain extracts were prepared as described previously (Chung et al. 2002b; Uchida et al. 1988). Briefly, the whole brain of adult male Hooded Wistar rat ( $\sim 200$  g) was removed and homogenized in one volume of Hanks buffer, followed by centrifugation at 100g. The supernatant was collected and filter-sterilized (0.22 µm filter, Millipore, Billerica, MA,

USA). The total protein concentration of freshly prepared rat brain extracts was determined by the Bradford method (Bradford 1976). Extracts were used immediately at a total protein concentration of  $150 \, \mu g \, \text{ml}^{-1}$  (Chung et al. 2002b).

## Culture of cerebral cortical cells

Cultures of cerebral cortical cells were prepared as described previously with slight modifications (Chung et al. 2002b; Sewell et al. 1995). Briefly, cerebral cortices of Wistar rat fetuses (day 15) were removed and cells were dissociated mechanically by passing coarsely minced cortical tissues gently and repeatedly through a pipette. Tissue debris was removed by gently passing through a filter. Cells were washed and suspended in a specific culture medium developed for selective neuronal growth, consisting of Neurobasal-A medium (GIBCO), 0.1% (f/c) B-27 supplement (GIBCO), and 0.1 mM (f/c) L-glutamine (sigma). Then, cells were seeded on polylysine-coated 24- and 96-well culture plates at a density of  $5 \times 10^5$  and  $2 \times 10^6$  cells/ml, respectively. Cultures were maintained in a 37°C chamber under an atmosphere of humidified air containing 5% CO<sub>2</sub>. Four hours later, the culture medium was replaced with fresh medium containing 150 μg ml<sup>-1</sup> rat brain extract and 10–20 μg ml<sup>-1</sup> proteins (Chung et al. 2002b). Cultures were maintained in the same conditions for 3 days.

## Calculations and analysis of neuron culture

To quantitatively analyze effect of hGIF and its mutants on initial neurite extension, cultures (24-well) were fixed in 4% paraformaldehyde and then stained with Trypan blue. Neurite length was obtained by measuring the distance between the end of the neurite and cell surface using the Leica Qwin program (Rueil Malmaison, France), and at least 60 neurons with more than 100 neurites were recorded in each individual experiment.

The percentage of neuron survival was determined by a MTT cell reduction assay (Cory et al. 1991). Briefly, 20 µl of MTT solution (5 mg ml<sup>-1</sup>) in 0.02 M PBS was added to each well (96-well plate) and incubated at 37°C for 3 h. After incubation, unreacted dye and medium were removed, and 150 µl dimethyl sulfoxide (DMSO) was added to each well



to solubilize the MTT-formazan product formed. Then, the MTT-formazan products were quantified by measuring the absorbance at 490 nm using a Bio-Rad 680 Multireader (California, USA), and background readings of MTT incubated in cell-free medium were subtracted from each value before calculations.

## Results and discussion

After expression and purification, the yield of each mutant protein was about  $8 \sim 9$  mg per liter culture. Their molecular masses were determined by ESI-MS and were in good agreement with the theoretical values (Table 1). The metal contents of hGIF and its variants were obtained by flame atomic absorption spectrophotometer (Table 1). It was demonstrated that  $Cd^{2+}$  is a useful probe for  $Zn^{2+}$  binding sites in

Table 1 Mass spectrometry results and metal contents of hGIF and its mutants

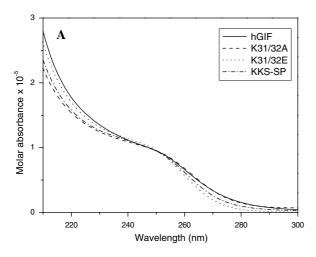
Protein	Mr (calc.) <sup>a</sup> (Da)	Mr (meas.) <sup>b</sup> (Da)	Metal contents
hGIF	7,071	7,070.7	$7 \pm 0.3$
K31/32A	6,957	6,956.8	$7 \pm 0.2$
K31/32E	7,073	7,072.7	$7 \pm 0.4$
KKS-SP	6,912	6,910.0	$7 \pm 0.2$

<sup>&</sup>lt;sup>a</sup> The theoretical molecular mass

MTs, providing a wealth of structural information on Zn<sup>2+</sup>-substituted MTs (Vasak 1998). Furthermore, Cd<sup>2+</sup>-substituted MTs show a number of advantages over Zn<sup>2+</sup>-substituted MTs, including a tendency toward higher oxidation stability, a pronounced and thus easier to detect and quantify LMCT band, as well as the use of the <sup>111</sup>Cd or <sup>113</sup>Cd isotopes for NMR experiments, etc. Thus, Cd<sup>2+</sup> was frequently used as a substitute of Zn<sup>2+</sup> for structural studies of hGIF (Cai et al. 2006; Hasler et al. 2000; Romero-Isart et al. 2002). In the present studies, Cd<sup>2+</sup>-substituted MTs were adopted in both spectroscopic and metal-binding studies while Zn<sup>2+</sup>-substituted MTs were only used in bioassay because Cd<sup>2+</sup>-MT is cytotoxic to neurons (Sewell et al. 1995).

Figure 2a shows the UV-vis spectra of Cd<sub>7</sub>-hGIF and its mutants. All the spectra maintained similar shape and showed absorption shoulders around 250 nm, characteristic of Cd–S ligand-to-metal charge transfer (LMCT) (Vasak et al. 1981). There was no absorption around 280 nm due to the absence of aromatic amino acid residue.

Circular dichroism spectroscopy is commonly adopted to investigate the metal–thiolate cluster geometry and secondary structures of the enfolding polypeptide chains of MTs (Cai et al. 2006; Faller and Vasak 1997; Hasler et al. 1998; Kagi and Schaffer 1988; Romero-Isart et al. 2002; Stillman et al. 1987). Figure 2b shows the CD spectra of Cd<sub>7</sub>-hGIF and its mutants in the range of 200–300 nm. The CD spectrum of hGIF nicely matched the results



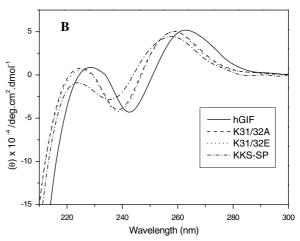


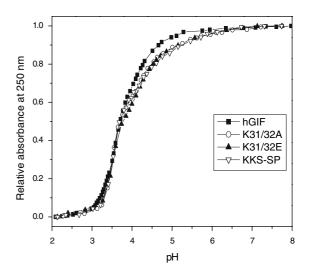
Fig. 2 UV (a) and CD (b) spectra of Cd<sub>7</sub>-hGIF, the K31/32A mutant, the K31/32E mutant and the KKS-SP mutant. All proteins are dissolved in 10 mM Tris-HCl, 100 mM KCl, pH 8.0



<sup>&</sup>lt;sup>b</sup> The measured molecular mass

from previous reports (Hasler et al. 1998, 2000) and showed bands at 259 (+), 239 (-) and 225 (+) nm. The bands around 260 (+) and 240 nm (-) are the contributions from Cd-thiolate clusters and the band around 225 nm (+) has been assigned to the interaction between the Cd-thiolate clusters and the peptide wrapping around (Stillman et al. 1987). The CD spectra of the K31/32A and K31/32E mutants were almost the same as that of hGIF, indicating that change of the positive charge of lysine residues did not affect the overall structure and metal-thiolate cluster geometry apparently. However, compared to hGIF, the CD profile of the KKS-SP mutant was blue shifted, which undoubtedly suggested alterations of the cluster geometry and conformation of enfolded polypeptide in this mutant. It was demonstrated that the two domains in metallothionein do not work independently and domain-domain interactions do exist in MT and affect the structure and function of each domain (Ding et al. 2007; Jiang et al. 2000; Zangger and Armitage 2002). In the KKS-SP mutant, introduction of a rigid proline residue in the short linker region may alter the relative orientation of the two domains and changed the domain-domain interaction of hGIF to some extent, thus impacting the metal-thiolate geometry and polypeptide structure.

Ultraviolet absorbance at 250 nm was recorded during pH titration in order to investigate the stability of the Cd-thiolate clusters (Fig. 3). Displacement of Cd<sup>2+</sup> from MTs with increasing concentrations of proton was followed by a decrease in the Cd-S LMCT absorption band at 250 nm. According to previous studies, pH titration curves of MTs are usually divided into two stages reflecting different stabilities of the two metal-thiolate clusters and the metal-thiolate cluster in the  $\beta$ -domain is more easily dissociated (Cismowski and Huang 1991). Stabilities of the Cd-thiolate clusters were quantitatively studied by calculating their apparent Cd-binding constants at pH 7.0, which could be easily estimated from the midpoint values of pH titration curves (Kagi and Vallee 1961; Shaw et al. 1991; Vasak and Kagi 1983; Wang et al. 1994). Compared with hGIF, the stabilities of the  $Cd_3S_9$  cluster in the  $\beta$ -domain of the mutant proteins significantly decreased (Table 2). This result undoubtedly reflected that the linker <sup>31</sup>KKS<sup>33</sup> was more helpful in maintaining the stability of the metal-thiolate in the  $\beta$ -domain of hGIF. Recently, we predicted the structure of hGIF by



**Fig. 3** UV absorbance at 250 nm versus pH during pH titration of Cd<sub>7</sub>-hGIF (*solid squares*), the K31/32A mutant (*open circles*), the K31/32E mutant (*solid triangles*) and the KKS-SP mutant (*open triangles*). All proteins are dissolved in 10 mM Tris–HCl, 100 mM KCl, pH 8.0

molecular dynamics simulation and found Lys32 in the linker formed a hydrogen bond with Cys22 (2.92 Å) in the  $\beta$ -domain [Our unpublished data]. Interestingly, the same hydrogen bond between Lys31 (according to Lys32 in hGIF) and Cys21 (according to Cys22 in hGIF) is also observed in the X-ray structure of rat liver MT2. This hydrogen bond is believed to play an important role in the stability of metal–thiolate cluster in the  $\beta$ -domain of MT2 (Braun et al. 1992; Robbins et al. 1991). Hence, when we changed the KKS to other sequence, the hydrogen bond between Lys32 and Cys22 might vanish, thus decreased the stability of the metal–thiolate clusters in the  $\beta$ -domain of hGIF.

The reaction of metallothionein with DTNB is used to estimate the solvent accessibility of the metal–thiolate clusters and usually exhibits biphasic kinetics (Shaw et al. 1991). In the case of hGIF, the fast and slow phase correspond to the reaction of DTNB with the  $\beta$ -domain and  $\alpha$ -domain, respectively (Sewell et al. 1995). Observed rate constants were obtained by plotting  $\ln(A_{\infty}-A_t)$  versus time and these constants were listed in Table 3. The fast reaction rate constants ( $k_f$ ) of the three mutants were similar and were about twice that of hGIF, showing that the solvent accessibilities of the Cd<sub>3</sub>S<sub>9</sub> clusters in the three mutants were greatly enhanced. By contrast, the slow reaction rate constants ( $k_s$ ) of the three



Protein	Midpoint (Cd <sub>3</sub> S <sub>9</sub> )	$K_{\text{Cd,app}} (M^{-1})$ of the $\beta$ -domain (×10 <sup>13</sup> )	Midpoint (Cd <sub>4</sub> S <sub>11</sub> )	$K_{\text{Cd,app}} (M^{-1}) \text{ of the } $ $\alpha\text{-domain } (\times 10^{15})$
hGIF	4.08	7.68	3.31	1.32
K31/32A	4.37	1.30	3.33	1.44
K31/32E	4.29	1.03	3.36	1.02
KKS-SP	4.39	0.91	3.35	1.09

Table 2 Apparent Cd-binding constants of hGIF and its mutants

Apparent Cd-binding constants ( $K_{\text{Cd, app}}$ ) were determined by the competition between metal ions and protons for the thiolate ligands and extrapolated to pH 7.0. For details, please see "Materials and methods"

**Table 3** Observed rate constants for the reaction of DTNB with hGIF and its mutants

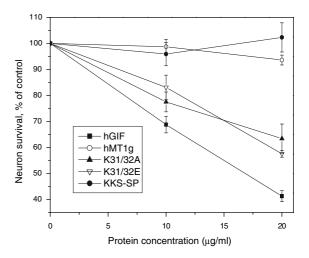
Protein	$k_{\rm f}^* \ (\times 10^{-3} \ {\rm s}^{-1})$	$k_{\rm s}^* \ (\times 10^{-4} \ {\rm s}^{-1})$
hGIF	$2.5 \pm 0.1$	$9.3 \pm 0.2$
K31/32A	$4.2 \pm 0.1$	$8.6 \pm 0.1$
K31/32E	$4.3 \pm 0.1$	$7.5 \pm 0.2$
KKS-SP	$4.3 \pm 0.1$	$8.5 \pm 0.2$
THIS SI	1.5 ± 0.1	0.5 ± 0.2

 $k_{\rm f}$  is the rate constant for the fast reaction,  $k_{\rm s}$  is the rate constant for the slow reaction

mutants were similar to that of hGIF, indicating the solvent accessibilities of the Cd<sub>4</sub>S<sub>11</sub> clusters in the three mutants did not change much.

Figure 4 shows the effect of the hMT1g (negative control), hGIF (positive control) and its mutants on neuron survival determined by a MTT cell reduction assay. MTT is reduced to blue formazan by mitochondrial dehydrogenase of living cells but not by dead cells (Cory et al. 1991). Apparently, hGIF revealed the neuronal growth inhibitory activity, which was not shared by hMT1g. This result agreed well with those reported previously (Chung et al. 2002b; Hasler et al. 2000; Sewell et al. 1995). Compared to hGIF, the K31/32A and K31/32E mutants exhibited similar but significantly reduced bioactivity, while the KKS-SP mutant lost bioactivity completely. Hence, we concluded that the conserved liker KKS played important roles in the bioactivity of hGIF.

Effect of hGIF and its mutants on neurite extension was also studied (Fig. 5). The average neurite length of neurons treated by hMT1g, hGIF, the K31/32A mutant, the K31/32E mutant and the KKS-SP mutant were  $164 \pm 4$ ,  $85 \pm 3$ ,  $115 \pm 3$ ,  $127 \pm 5$  and  $159 \pm 4$  µm, respectively. This result paralleled well with the data of neuron survival mentioned above.



**Fig. 4** Effect of  $Zn_7$ -hGIF (solid squares), the K31/32A mutant (solid triangles), the K31/32E mutant (open triangles) and the KKS-SP mutant (solid circles) on survival of cortical neurons cultured in the presence of adult rat brain extract (150  $\mu$ g/ml) after 3 days, determined by a MTT cell reduction assays. As a comparison, result of hMT1g (open circles) under the same conditions was also shown. *Error bars* represent standard error values

The neuronal growth inhibitory activity of hGIF is unique in MT family (Palmiter et al. 1992; Uchida et al. 1991; Uchida and Tomonaga 1989). More and more evidences revealed that the bioactivity of hGIF mainly relates to the essential metal release (Erickson et al. 1997; Palmiter 1995) and its characteristic conformation (Ni et al. 2007; Sewell et al. 1995), especially the  $\beta$ -domain. The metal-releasing ability, determined by the stability and solvent accessibility of the metal-thiolate cluster in the  $\beta$ -domain of hGIF, is exquisitely modulated by domain-domain interaction and protein's conformation (Ding et al. 2008, 2007; Faller et al. 1999; Ni et al. 2007). Furthermore, the characteristic conformation of hGIF is also



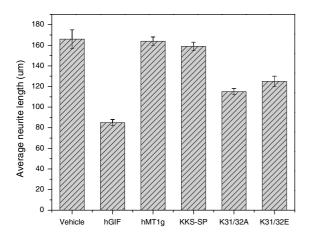
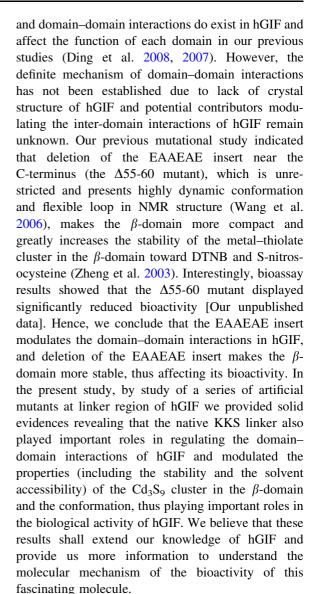


Fig. 5 Effect of  $Zn_7$ -hGIF, the K31/32A mutant, the K31/32E mutant and the KKS-SP mutant on neurite extension in the presence of adult rat brain extract (150  $\mu$ g/ml) after 3 days, determined by average neurite length of 60 neurons. As a comparison, result of hMT1g (negative control) and 0.1% PBS (vehicle) under the same conditions was also shown. *Error bars* represent standard error values

suggested to provide an interface for the interaction with its partner protein (Ni et al. 2007; Sewell et al. 1995). In the present study, it was found that both the K31/32A and K31/32E mutants showed reduced bioactivity. Based on the results of biochemical studies, the overall structure of the two mutants did not change much, but the stability and solvent accessibility of the Cd<sub>3</sub>S<sub>9</sub> clusters of the two mutants altered much. Thereby, the reductions in bioactivity of the two mutants were most likely due to the change in the stability and solvent accessibility in the Cd<sub>3</sub>S<sub>9</sub> cluster. Interestingly, the bioactivity of the KKS-SP mutant was completely abolished although the stability and solvent accessibility of the KKS-SP mutant were similar to those of the K31/32A and K31/32E mutants. However, it should be noted that mutating KKS to SP not only changed the stability and solvent accessibility of metal-thiolate cluster in the  $\beta$ domain, but more significantly altered the domaindomain interactions/relative orientation of the two domains of hGIF, thus making the KKS-SP mutant lose its bioactivity completely.

#### **Conclusions**

It has been demonstrated that the two domains of hGIF do not work independently (Faller et al. 1999)



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